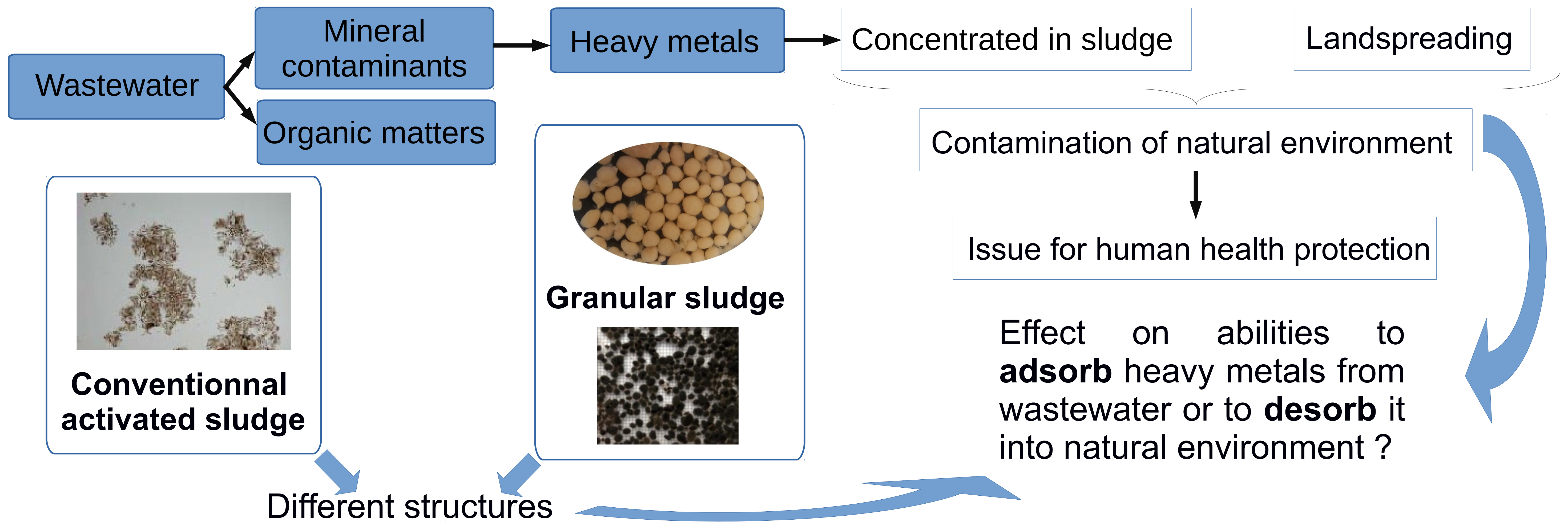


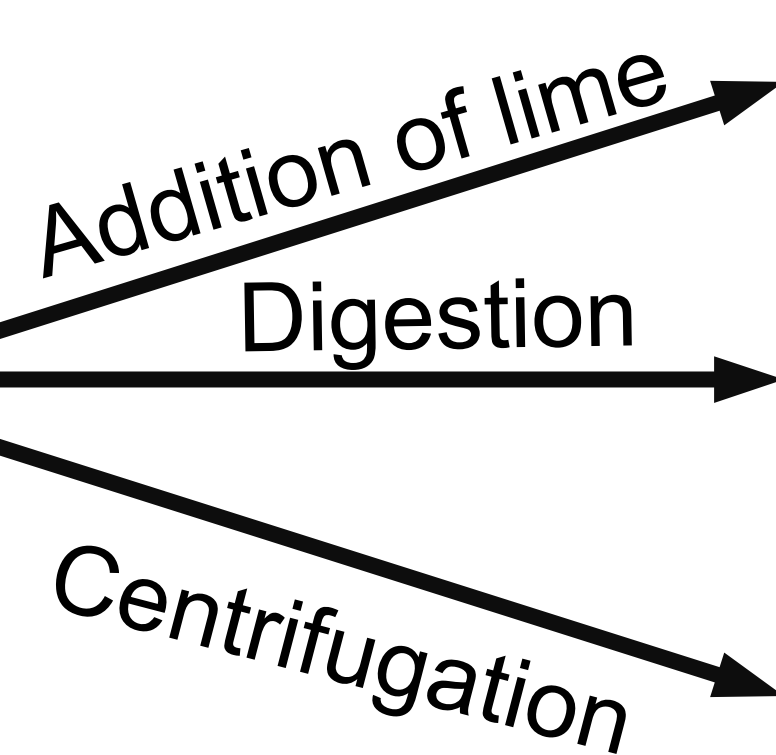
Introduction



Methods

Sludge samples

CAS : Conventional activated sludge (WWTP of Limoges)



BCh
BD
BC

BAN : Anaerobic granular sludge (Industrial treatment plant)

BAE : Aerobic granular sludge (Industrial treatment plant)

Analysis

- Proportion and nature of surface functional groups by pKa measurements (Laurent et al., 2011),
- Copper distribution in the native matrix with a modified BCR protocol (Kazi et al., 2006) and Copper concentrations by Flame Atomic Absorption Spectroscopy,
- Copper adsorption isotherms by adding copper (concentrations from 0 mg/L to 400 mg/L),
- Desorption ability of sludge by re-suspending sludge in NaCl solution (L/S ratio of 50 L/kg).

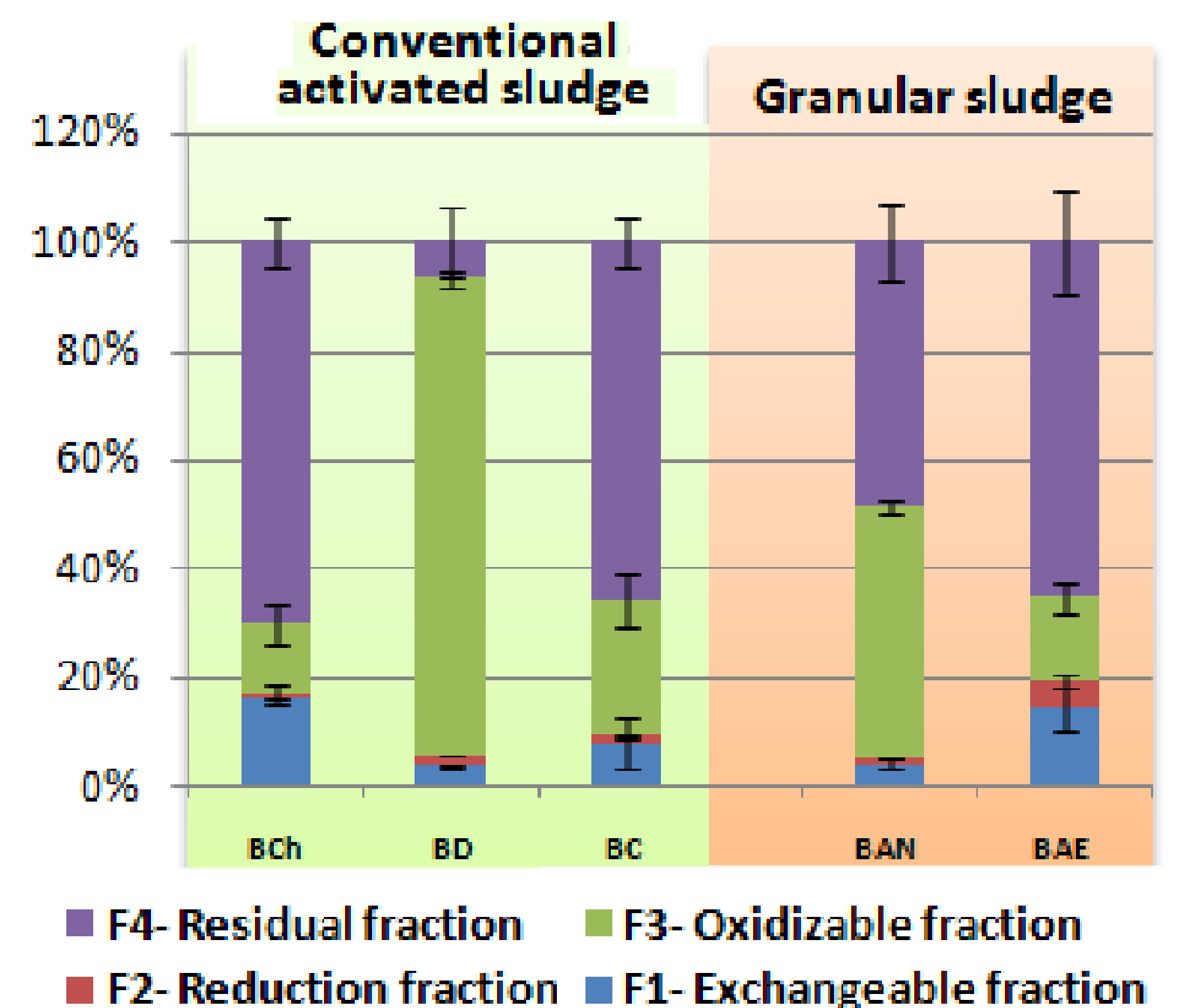
Results

Sludge	Conventional Sludge			Granular Sludge		
	BCh	BD	BC	BAN	BAE	
pH	13,1	7,5	5,9	6,8	8,7	
Cu Adsorption K_d (L/g)	$0,098 \pm 0,022$	$0,324 \pm 0,077$	$0,277 \pm 0,059$	$0,134 \pm 0,026$	$0,731 \pm 0,093$	
% Cu desorption ability	4,8	1,4	1,7	0,4	3,4	
Particulate fractions	Carboxyl	52%	81%	61%	58%	65%
	Phosphoric	33%	9%	20%	26%	17%
	Amine Hydroxyde	15%	10%	19%	16%	18%

Copper adsorption capacity : BAE > BD > BC > BAN > BCh

Copper desorption percentage : BCh > BC > BD > BAE > BAN

Proportion of carboxyl fonction : BD > BAE ≈ BC ≈ BAN > BCh



Exchangeable fraction : BCh ≈ BAE > BC > BAN ≈ BD

Conclusion

High exchangeable fraction →

High copper desorption

High exchangeable fraction + High carboxyl groups →

High copper adsorption

Reference

Laurent, J., Casellas, M., Carrère, H., Dagot, C. (2011). Effects of thermal hydrolysis on activated sludge solubilization, surface properties and heavy metals biosorption. *Chem. Eng. J.* **166** (3), 841–849.

Kazi, T.G., Jamali, M.K., Siddiqui, A., Kazi, G.H., Arain, M.B., Afridi, H.I. (2006). An ultrasonic assisted extraction method to release heavy metals from untreated sewage sludge samples. *Chemosphere* **63** (3), 411–420